

# Effects of different carriers for adsorption of $^{125}\text{I}$ on brachytherapy sources

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**Background:** One of the key techniques for the preparation of  $^{125}\text{I}$  seeds is adsorption of  $^{125}\text{I}$  onto silver bits coated by palladium (pd). Carriers played an important role in the adsorption of  $^{125}\text{I}$  on palladium. KI is used as a carrier for fixing of  $^{125}\text{I}$  onto silver wire bits coated with palladium. **Materials and Methods:** Three procedures KI, KOH,  $\text{NH}_4\text{OH}$  were investigated for adsorption of  $^{125}\text{I}$  activity by different carriers. **Results:** Adsorption percentage of  $^{125}\text{I}$  on treated silver beads showed about 85% by using KI as a carrier, the KOH showed 74% and 65% for  $\text{NH}_4\text{OH}$ . **Conclusion:** The results indicated that, the use of KI as a carrier was suitable for adsorption of  $^{125}\text{I}$  on treated silver seeds with Pd than KOH and  $\text{NH}_4\text{OH}$ . *Iran. J. Radiat. Res.*, 2012; 10(2): 105-107

**Keywords:** I-125, brachytherapy, seed source.

## INTRODUCTION

Iodine-125 brachytherapy sources are being used for interstitial brachytherapy implants in various tumor sites and particularly for prostatic carcinomas. Patients with malignant tumors continue to have a poor prognosis despite treatment with surgery, radiation therapy and chemotherapy. Although external beam radiation is quite effective in tumor control, the treatment is limited by the incidence of radiation injury in the surrounding normal tissue (1, 2). By implanting the radiation source directly into the tumor, a relatively high dose can be delivered to the tumor, while the surrounding tissue receives a much more modest dose (3-7).

Brachytherapy involves the delivery of high doses of radiation to the tumor while sparing normal surrounding tissue using temporarily or permanently implanted radioactive sources (8). While interstitial I-125 brachytherapy appears to be beneficial, the risk of radiation necrosis and the need

for reoperation for necrosis has prompted investigation of other options to reduce or limit the radiation dose while enhancing tumor control (8).

Almost all stereo tactic implants are made with either I-125 or Ir-192 sources. Of the two, commonly used radio nuclides I-125 is the more popular choice. Its low-photon energy makes shielding easy, and tends to reduce normal tissue dose with respect to tumor dose.

Implants with I-125 seeds, because of lower energies and greater absorption in tissue, require much higher source strength than those indicated in the classical tables (for the same implant area).

Iodine-125 fixed on silver bars make a source core which, after being sealed in titanium capsules, are recognized as the so-called seed sources used in radiation therapy for treatment of intraocular and brain tumors, and as permanent implants for prostate cancer (9-10). The commercial manufacturers apply different methods for preparing source cores such as, adsorption on organic materials, ion-exchange resins, ceramic beads, palladium bars, silver coated beads, etc. Although some results of investigations concerning fixing of iodine-125 on silver are reported (11-13).

## MATERIALS AND METHODS

$^{125}\text{I}$  as sodium iodide in dilute sodium hydroxide solution was obtained from MDS Nordions. Silver wire of 0.5mm ( $\emptyset$ ) with pt. 99.9% was obtained from local suppliers,

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and it was cut to 3 mm length mechanically. Palladous chloride, KI, KOH and NH<sub>4</sub>OH were obtained from Merck-Schuchardt, 8011 Hohenbrunn bei munchen.

#### **Pretreatment of silver wires**

The silver wire was cut mechanically to a length of 3 mm. Twenty five pieces of such wires were selected. The wires were washed with acetone and then with hot and cold water 3 times. The washed wires were treated with HCl 3M and subsequently were washed with acetone, and then with hot and cold water. The wires were completely dried under an IR lamp. The silver beads were immersed in aqueous palladous chloride solution (0.5mg/ml) 0.03 M at pH 5.5-6.5. The solution was gradually heated to 100°C and maintained at that temperature for 5 minutes and cooled in the room temperature. The wires were washed with acetone and distilled hot and cold water three times and dried and they were stored in sealed vials under normal atmospheric pressure.

#### **Adsorption of <sup>125</sup>I on silver wires**

Treated silver wires were immersed in <sup>125</sup>I solution (30 mCi/ml) at alkaline pH. The solution was heated in different times for the different volume of <sup>125</sup>I solution and for the different activities.

The result of <sup>125</sup>I coating using KI as carrier was indicated that solubility of Pd into KI solution is an important parameter. Therefore, the other carriers can be used for <sup>125</sup>I coating on Pd coated silver wire such as KOH and NH<sub>4</sub>OH.

In these experiments, KI, KOH and NH<sub>4</sub>OH were used as carrier which adsorption of <sup>125</sup>I on Pd coated silver wire were founded 85%, 75% and 62%, respectively:

If KI was used as carrier, layer of <sup>125</sup>I would have PdI<sub>2</sub> form and for excess KI would have PdI<sub>4</sub> form. In case of using NH<sub>4</sub>OH as carrier, layer of <sup>125</sup>I would have PdI<sub>2</sub>(NH<sub>3</sub>)<sub>4</sub> form and excess NH<sub>4</sub>OH would have Pd(NH<sub>3</sub>)<sub>4</sub>I<sub>2</sub> form.

Treated silver wires with palladous chloride were immersed in 50 μL <sup>125</sup>I

solution accompaniment by 10 μL KI (0.03 M), 10 μL KOH (0.03 M) and 10 μL NH<sub>4</sub>OH (0.03 M) then heated at different temperature for various times.

#### **Cumulative preparations of cores**

Fifteen coated silver wires with Pd were taken and were subdivided into 3 groups of 5 pieces. Pieces of treated silver wires with palladium chloride in group of 5 were immersed cumulative in 160 μL <sup>125</sup>I solution accompaniment by 30μL, KI (0.03 M), 30μL KOH (0.03 M) and 30μL NH<sub>4</sub>OH (0.03 M) then heated at 60°C for 6 hours.

In these experiments, KI, KOH and NH<sub>4</sub>OH were used as carrier which adsorption of <sup>125</sup>I on Pd coated silver wire were found 93%, 89% and 85% respectively.

## **RESULTS AND DISCUSSION**

The result of <sup>125</sup>I coating, using KI as carrier, indicated that solubility of Pd into KI solution has been an important parameter. Therefore the other carriers can be used for <sup>125</sup>I coating on Pd coated silver wire such as KOH and NH<sub>4</sub>OH.

In these experiments, KI, KOH and NH<sub>4</sub>OH were used as carrier which adsorption of <sup>125</sup>I on Pd coated silver wire were found to be 85%, 74% and 62% (table 1).

**Table 1.** Comparison of adsorption of <sup>125</sup>I by different carrier.

Carrier	Activity range (mCi)	Percentage adsorption
KI	0.70 - 2.08	85%
KOH	0.85 - 2.03	74%
NH <sub>4</sub> OH	1.02 - 2.63	65%

The present study showed that, superiority of using KI as carrier in low activity experiments.

By using KOH as a carrier, 15 mCi of 25 mCi (60%) activity of <sup>125</sup>I on silver seeds coated with palladium was adsorbed.

When some pieces of treated silver wires with palladium immersed in KI, KOH and NH<sub>4</sub>OH; adsorption of <sup>125</sup>I were found to be

93%, 89% and 85% respectively (table 2).

Carriers played an important role in the adsorption of <sup>125</sup>I on the palladium coated silver wires. The use of KI as carrier has been studied by Mathew *et al.* (11).

The results indicated that, the use of KI as a carrier was more effective for adsorption of <sup>125</sup>I on treated silver seeds with Pd than KOH and NH<sub>4</sub>OH. It was found that, the percentage of adsorption was dependent on surface area (tables 1 and 2).

**Table 2.** Comparison of adsorption of <sup>125</sup>I by different carriers when some pieces of treated wires were immersed.

Carrier	Solution activity (mCi)	Seeds activity (μCi)	Percentage adsorption
KI	1.688	S <sub>1</sub> = 325 S <sub>2</sub> = 319 S <sub>3</sub> = 294 S <sub>4</sub> = 328 S <sub>5</sub> = 310	93%
KOH	1.719	S' <sub>1</sub> = 319 S' <sub>2</sub> = 308 S' <sub>3</sub> = 278 S' <sub>4</sub> = 321 S' <sub>5</sub> = 302	89%
NH <sub>4</sub> OH	1.528	S'' <sub>1</sub> = 366 S'' <sub>2</sub> = 342 S'' <sub>3</sub> = 354 S'' <sub>4</sub> = 359 S'' <sub>5</sub> = 343	85%

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